



*The Authority in Potassium and Magnesium*

K+S KALI GmbH

---

Cations in Ecuadorian tropical soils: different extraction methods and interpretation of the analysis

Dr. Bernd Ditschar



- **Calcium, Magnesium and Potassium as cations are fixed on organic matter and clay particles**
- **This fraction is exchangeable and always in balance with nutrients in the soil solution**
- **Soil analysis should analyze these two fractions – concentration in the soil solution and buffer capacity**



- **Weak extractants such as water or sodium bicarbonate reflect mainly the concentration in the soil solution, whereas strong extractants primarily indicate the capacity of the soil to supply nutrients to the soil solution (buffer capacity); Marschner and Rengel (2012)**
- **strong extractants measure additional cations from the soil => cations which are not interchangeable (e.g. in the case of Mg and Ca free carbonates)**



**CEC = Adsorption and desorption of cations**

**Two groups:**

**Acid forming group**

**H<sup>+</sup> and Al<sup>3+</sup>**

**Basic-forming group**

**Mg<sup>2+</sup>, Ca<sup>2+</sup>, K<sup>+</sup> and Na<sup>+</sup>**



## Cation-Relation: Why it is important in Latin America

- Information from soil testing is a guideline for fertilizer recommendations

- there are two methods:

**a) SLAN** – sufficient level of available nutrients (optimum level in the soil, it is plant specific)

**b) BCSR** – mineral balancing in soils (Albrecht papers)

| Cation relation in soil |    |
|-------------------------|----|
| Ca                      | 60 |
| Mg                      | 30 |
| K                       | 10 |

Ancupa (2013) reports an optimal cation relation in soils specific for oil palm

„Yellowing“ and „Drying“ of leaves in oil palm



**60:30:10**



**80:15:5**

**Ca:Mg:K**

## Example from Ecuador (Olsen „Modified“)

**Soil analysis:**

**organic matter 9.1%, sand 48%, lime 43%, clay 9%**

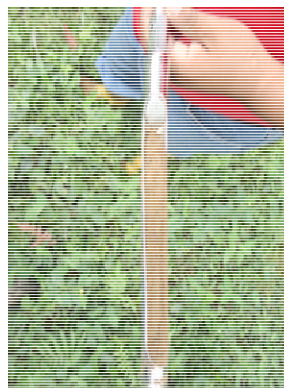
|        | Mg    | Ca    | K     | Total |
|--------|-------|-------|-------|-------|
| mg/ kg | 163   | 1614  | 322   | 2099  |
|        | Medio | M-A   | Alto  |       |
|        | 7,76% | 76,9% | 15,3% |       |

**Interpretation (according to ancupa, Ecuador):**

**Strong Mg-deficiency due to high Ca content!**



# K+S KALI GmbH Example from Ecuador



## Soil sampling and Mg-deficiency in Oil plam

Source: own data





**It might be that there is a specific cation ratio which is favorable to plants, especially for oil palm.**

**If the extraction of cations from the exchange complex depends of the “strongness” of the extractant the total amount of Ca, Mg an K differs on the method used.**

**And**

**Are the extraction methods comparable?**



In Ecuador the “Olsen modified” method is the standard method for Ca, Mg and K but it is originally a method for phosphorous

Different methods for soil testing in Latin America:

| Extractant                          | Extraction time | Dissolution |
|-------------------------------------|-----------------|-------------|
| <b>Water</b>                        | <b>1 h</b>      | <b>1:5</b>  |
| NaCl 0.1M                           | 1 h             | 1:10        |
| KCl 0.1M                            | 1 h             | 1:10        |
| CaCl <sub>2</sub>                   | 1 h             | 1:10        |
| Mehlich-3                           | 15`             | 1:10        |
| Ammonium-Acetate 1M                 | 1 h             | 1:20        |
| Sodium-Acetate 1M                   | 1 h             | 1:20        |
| Olsen Modified (Olsen + EDTA 0.05M) | 30`             | 1:20        |

# K+S KALI GmbH

## Ecuadorian Soils analyzed

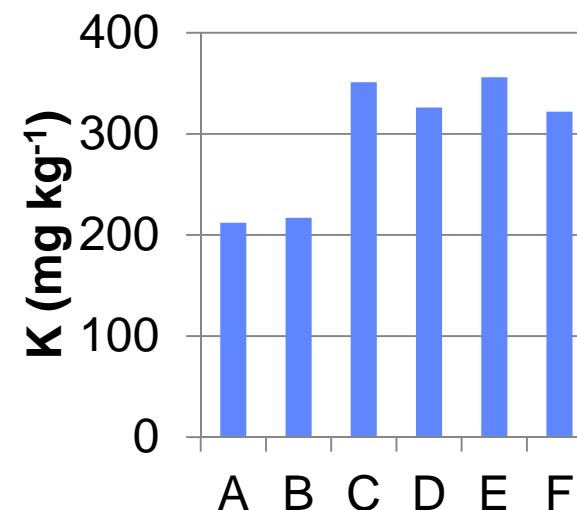
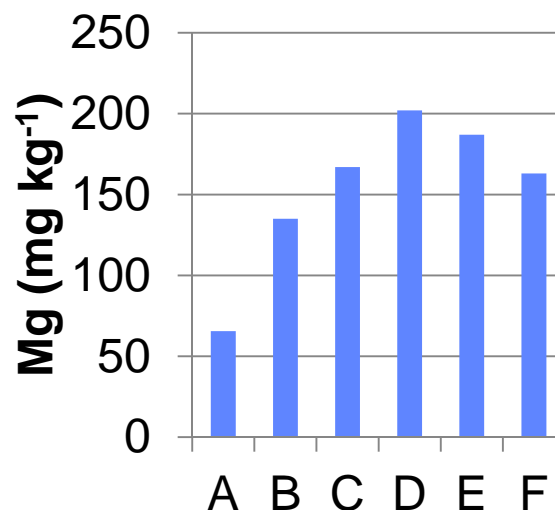
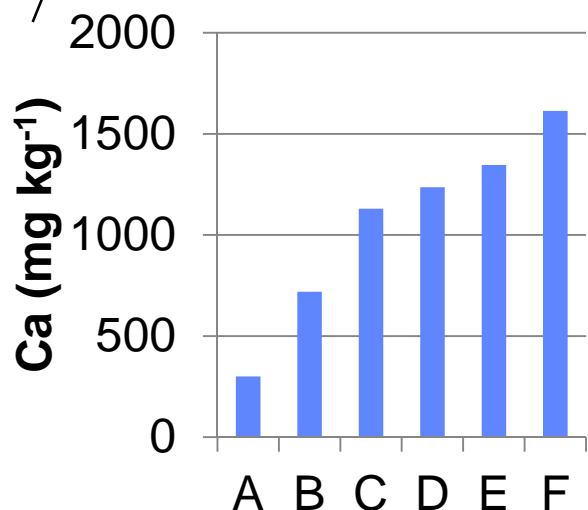


| Soil                    | Organic Matter (%) | Sand (%)  | Lime (%)  | Clay (%)  |
|-------------------------|--------------------|-----------|-----------|-----------|
| <b>Franca arcillosa</b> | <b>6.9</b>         | <b>28</b> | <b>32</b> | <b>40</b> |
| <b>Arcilla</b>          | <b>5.7</b>         | <b>23</b> | <b>30</b> | <b>47</b> |
| <b>Franca arenosa</b>   | <b>9.1</b>         | <b>48</b> | <b>43</b> | <b>9</b>  |
| <b>Franca</b>           | <b>7.7</b>         | <b>40</b> | <b>44</b> | <b>16</b> |
| <b>Franca arenosa</b>   | <b>10.8</b>        | <b>45</b> | <b>46</b> | <b>9</b>  |

Soil sampling: 13.8.2014 in Ecuador at different oil palm plantations

# K+S KALI GmbH

## Soil sample 3

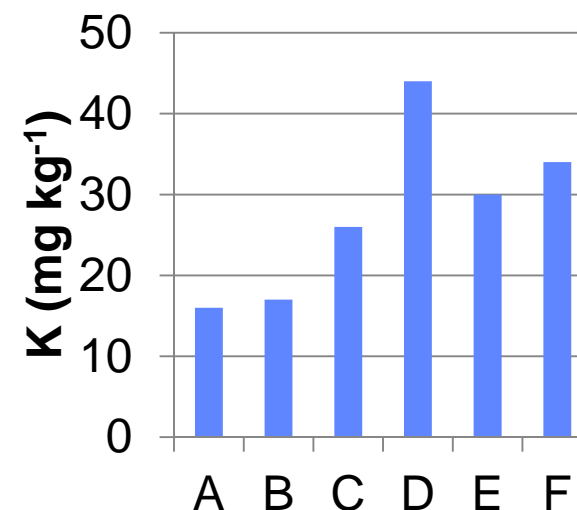
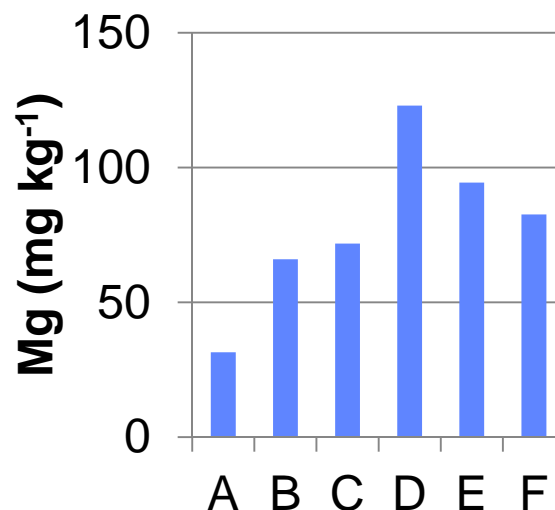
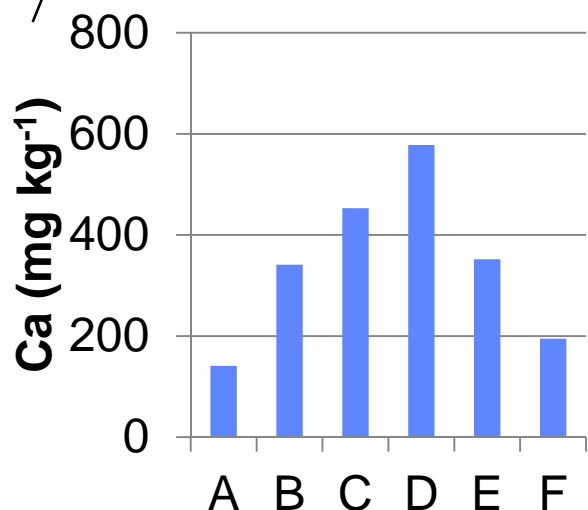


**A** (NaCl); **B** (KCl/CaCl<sub>2</sub>), **C** (Mehlich-3), **D** (Ammonium-Acetate), **E** (Sodium-Acetate) and **F** (Olsen Modified)

| Soil           | Organic Matter (%) | Sand (%) | Lime (%) | Clay (%) |
|----------------|--------------------|----------|----------|----------|
| Franca arenosa | 9.1                | 48       | 43       | 9        |

# K+S KALI GmbH

## Soil sample 5



**A** (NaCl); **B** (KCl/CaCl<sub>2</sub>), **C** (Mehlich-3), **D** (Ammonium-Acetate),  
**E** (Sodium-Acetate) and **F** (Olsen Modified)

| Soil           | Organic Matter (%) | Sand (%) | Lime (%) | Clay (%) |
|----------------|--------------------|----------|----------|----------|
| Franca arenosa | 10.8               | 45       | 46       | 9        |





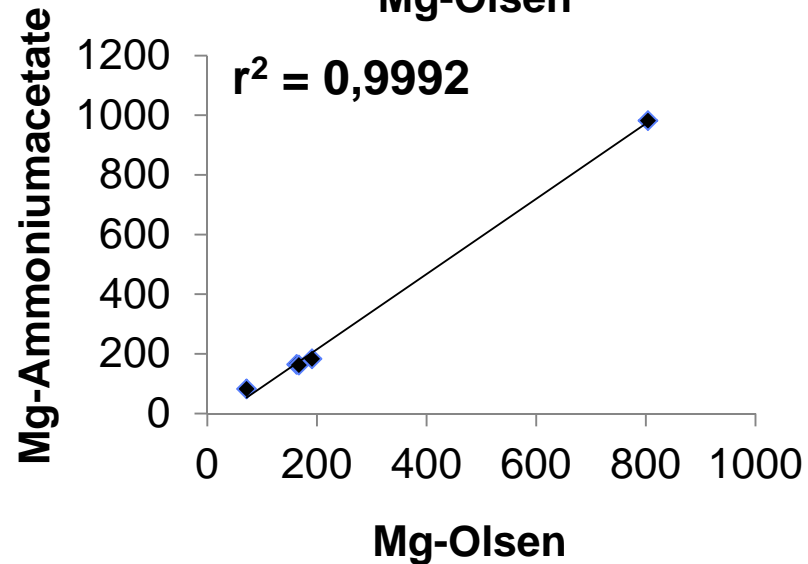
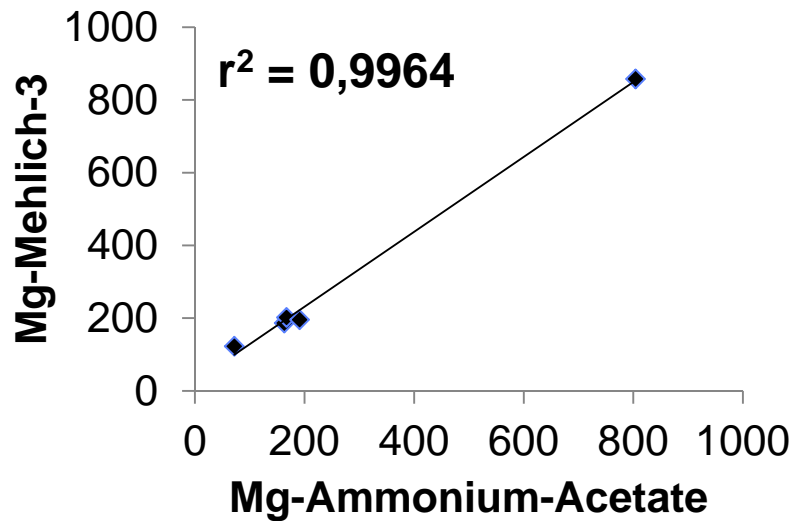
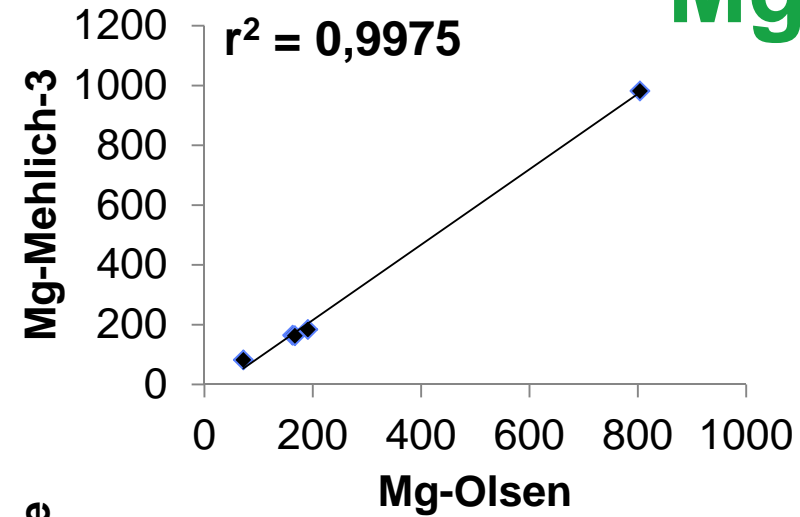
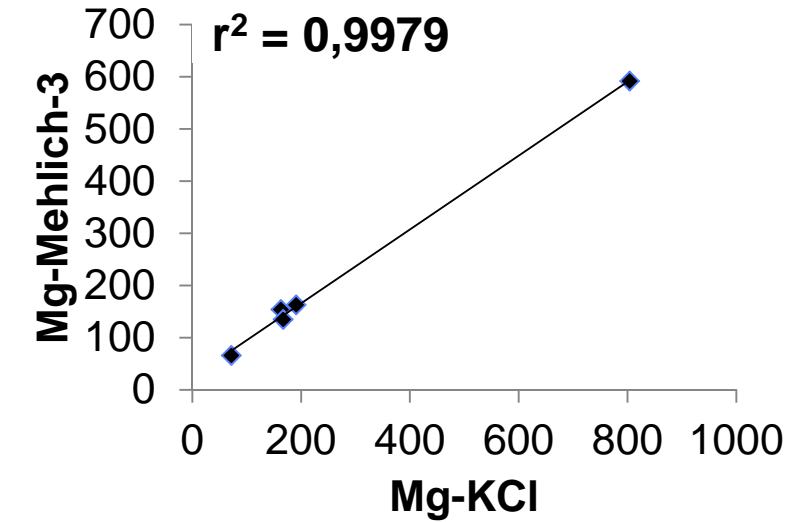
- The total amount of interchangeable cations of Ca, Mg and K ( $\text{mg kg}^{-1}$ ) found by each method are highly different
- Ammonium-Acetate showed the highest Mg levels in both soils; similar trend was found by the other methods
- Important is the interpretation of the analysis to give soil nutrient levels

**But**

**Are the different methods comparable?**

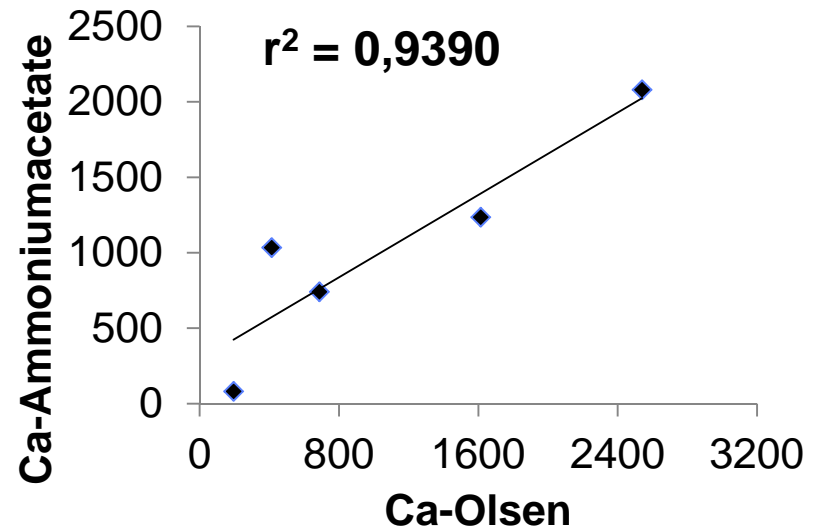
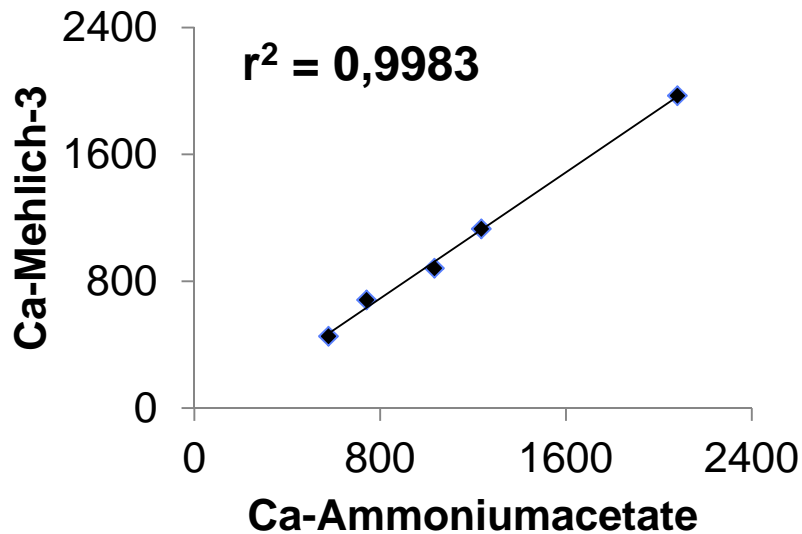
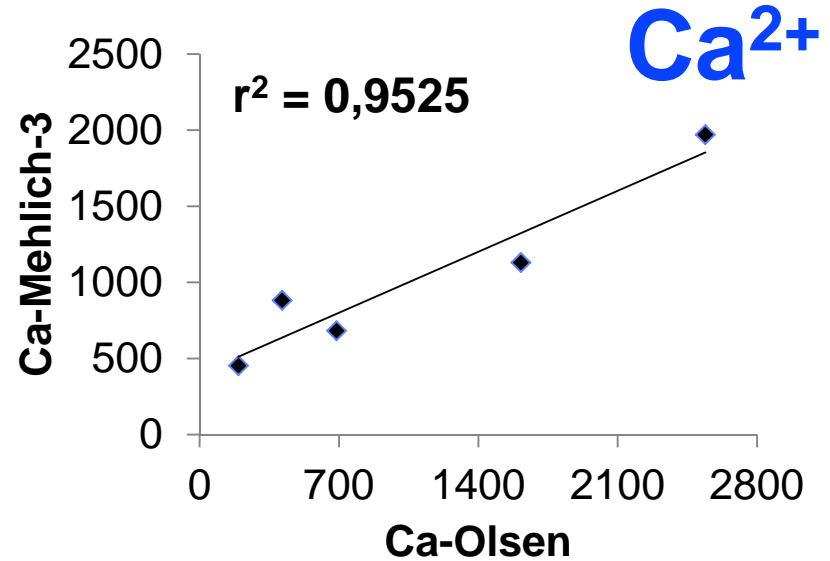
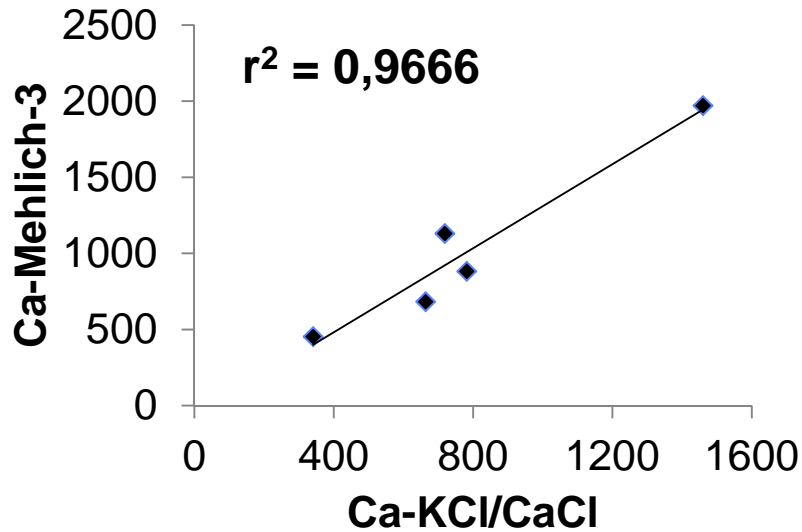


Mg<sup>2+</sup>



**Mg<sup>2+</sup>**

|                  | KCl    | NaCl   | Mehlich-3 | Ammonium-Acetate | Sodium-Acetate | Olsen "Modified" |
|------------------|--------|--------|-----------|------------------|----------------|------------------|
| KCl              | x      |        |           |                  |                |                  |
| NaCl             | 0.9917 | x      |           |                  |                |                  |
| Mehlich-3        | 0.9979 | 0.9839 | x         |                  |                |                  |
| Ammonium-Acetate | 0.9910 | 0.9681 | 0.9964    | x                |                |                  |
| Sodiumacetate    | 0.9944 | 0.9756 | 0.9989    | 0.9990           | x              |                  |
| Olsen Modified   | 0.9935 | 0.9713 | 0.9975    | 0.9992           | 0.9988         | x                |



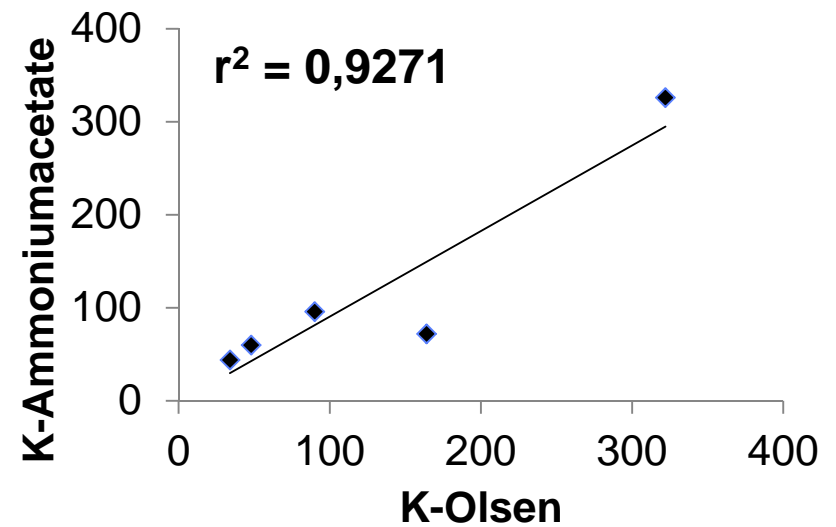
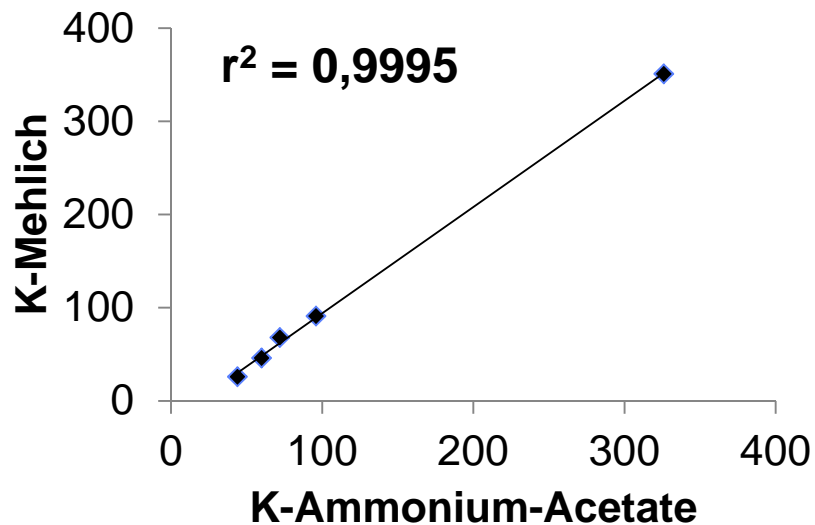
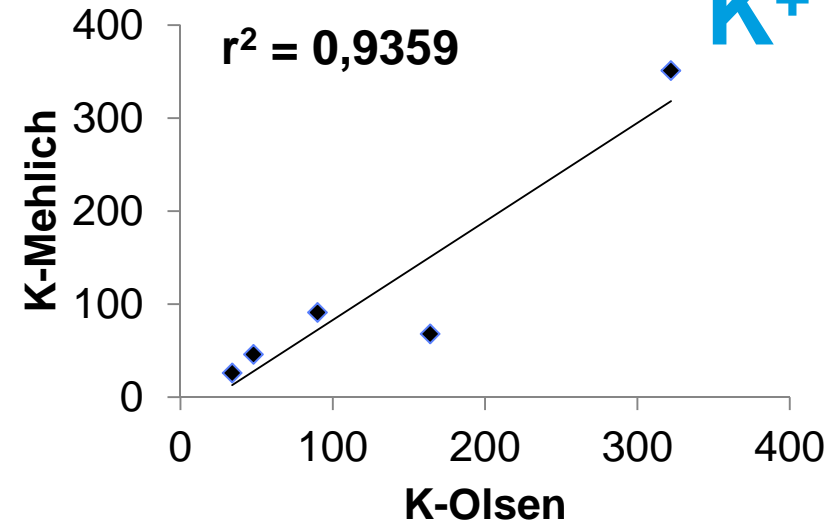
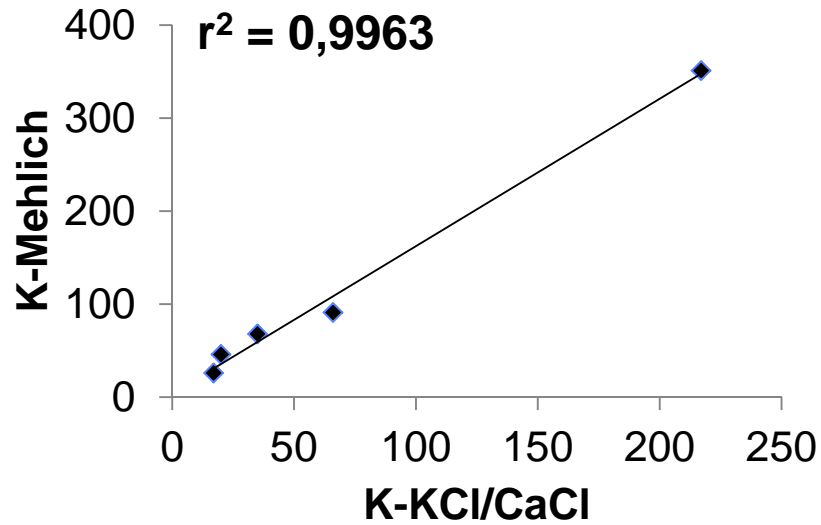
**Ca<sup>2+</sup>**

|                  | KCl    | NaCl   | Mehlich-3 | Ammonium-Acetate | Sodium-Acetate | Olsen „Modified“ |
|------------------|--------|--------|-----------|------------------|----------------|------------------|
| KCl              | x      |        |           |                  |                |                  |
| NaCl             | 0.9379 | x      |           |                  |                |                  |
| Mehlich-3        | 0.9666 | 0.9430 | x         |                  |                |                  |
| Ammonium-Acetate | 0.9097 | 0.9407 | 0.9983    | x                |                |                  |
| Sodiumacetate    | 0.8850 | 0.9106 | 0.9728    | 0.9691           | x              |                  |
| Olsen Modified   | 0.8720 | 0.8838 | 0.9525    | 0.9390           | 0.9845         | x                |





**K<sup>+</sup>**



|                 | KCl    | NaCl   | Mehlich | Ammonium-Acetate | Sodium-Acetate | Olsen „Modified“ |
|-----------------|--------|--------|---------|------------------|----------------|------------------|
| KCl             | x      |        |         |                  |                |                  |
| NaCl            | 0.9951 | x      |         |                  |                |                  |
| Mehlich         | 0.9963 | 0.9965 | x       |                  |                |                  |
| Ammoniumacetate | 0.9961 | 0.9978 | 0.9995  | x                |                |                  |
| Sodiumacetate   | 0.9965 | 0.9988 | 0.9991  | 0.9995           | x              |                  |
| Olsen Modified  | 0.9198 | 0.9196 | 0.9359  | 0.9271           | 0.9320         | x                |



For **Magnesium** all methods show a high significance to each other ( $r^2 = 0,98 - 0,99$ )

For **Calcium** only Ammonium-Acetate vs Mehlich-3 and Sodium-Acetate vs Olsen „Modified“ show a high significance ( $r^2 = 0,98 - 0,99$ )

For **Potassium** only the method Olsen „Modified“ is not significant to all other methods

# Relation between cations in soil solution and interchangeable part

---



**How is the nutrient availability with **water** in comparison with the interchangeable cations of each extraction method?**

**Is there a relation between cations in the soil solution and the interchangeable part?**

# Relation between cations in soil solution and interchangeable part



| Water versus     | Ca (r <sup>2</sup> ) | Mg (r <sup>2</sup> ) | K (r <sup>2</sup> ) |
|------------------|----------------------|----------------------|---------------------|
| NaCl             | 0,4444               | 0,4402               | 0,9724              |
| KCl              | 0,4303               | 0,4992               | 0,9856              |
| Mehlich 3        | 0,6182               | 0,5365               | 0,9692              |
| Ammonium-Acetate | 0,6436               | 0,5609               | 0,9679              |
| Sodium-Acetate   | 0,7064               | 0,5564               | 0,9715              |
| Olsen „Modified“ | 0,6210               | 0,5024               | 0,8944              |





- **Water shows poor correlations of measured Mg and Ca contents to all extraction methods, only the measured K showed a good correlation**
- **The extracted amount of cations differ from the method used**
- **For Magnesium all methods showed a high significant correlation, in the case of Calcium the correlation was not as good as Mg and K**



- **In the case of Mg the Ammonium-Acetate method showed the highest interchangeable nutrient content**
- **the nutrient level found was „very high“ compared to „low“ (with KCl and Olsen „Modified“)**
- **Nevertheless the Ammonium-Acetate method showed highly significant correlations to the other methods**
- **For fertilizer recommendations based on soil tests the interpretation of the results is very important and should analyzed carefully => a comparison of the leaf nutrient-level is recommended**

**The End**

---



**Thank you very much !**

## **Acknowledgement**

Thanks to Diego Monteros from FERMAGRI S.A. for providing me the soil samples analysed in this investigation